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$(AgSbTe_2)_{1-x}(Bi_2Te_3)_x$ -based thermoelectric device for low-grade heat recovery

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Keywords: AgSbTe ₂ Thermoelectrics Near room-temperature Solid solution	Near room-temperature thermoelectric materials have promising applications for recovering low-grade waste heat, but high-performance <i>p</i> -type thermoelectric candidates are quite limited if compared with <i>n</i> -type ones. It is thus important to design new <i>p</i> -type materials with superior thermoelectric performance. AgSbTe ₂ has received plenty of attention as a promising <i>p</i> -type material candidate due to its intrinsically low thermal conductivity, which is further decreased by introducing vacancies and substitutional point defects by alloying with Bi ₂ Te ₃ in this work. With the additional help of Cd substitution at the Sb site, the optimized carrier concentration leads to a peak <i>zT</i> value of 0.93 at 450 K for (AgSb _{0.98} Cd _{0.02} Te ₂) _{0.9} (Bi ₂ Te ₃) _{0.1} , and the corresponding single-leg device achieves a conversion efficiency of ~4.2 % at a temperature gradient ΔT of ~162 K. By further pairing with an <i>n</i> - type Ag ₂ Se leg, a conversion efficiency of ~1.8 % is realized at a ΔT of ~93 K for the obtained module, sug- gesting its potential applications in the low-grade heat recovery.

1. Introduction

Thermoelectric materials enable direct conversion between thermal and electrical energy and have been used in solid-state power generation and refrigeration [1,2]. According to the figure-of-merit $zT = S^2 T/\rho(\kappa_{\rm E}+\kappa_{\rm L})$, high-performance thermoelectric materials are required to have high Seebeck coefficients (*S*), low electrical resistivities (ρ), and low electronic ($\kappa_{\rm E}$) and lattice thermal conductivities ($\kappa_{\rm L}$). Besides optimizing thermoelectric material properties [3,4], improving fabricated device performance plays an important role as well. For example, reducing the thermal and electrical contact losses between the material and the electrodes [5,6] as well as rationally designing the device structure [7–9] can effectively improve the energy conversion efficiency.

 Bi_2Te_3 -based materials have been widely used in thermoelectric cooling and power generation because of their excellent thermoelectric properties parallel to the direction of the layers [10–14]. As the dominant commercial material system for near room-temperature applications, however, its layered structure induces low mechanical performance, and it is prone to cleavage along the layer stacking direction, which is usually undesirable. Other potential alternatives such as $Mg_3(Bi, Sb)_2$ [15,16], MgAgSb [17–19], and Ag_2Se [20,21] have been continuously investigated for their power generation and cooling performance with promising applications.

Numerous efforts have been focused on the exploration of new thermoelectric materials, preferably to address the drawbacks of bismuth telluride and obtain isotropic alternatives with excellent near room-temperature properties. It is widely recognized that solid solution or doping can lead to stable structure and improve thermoelectric properties [22–26]. Meanwhile, solid solutions of different material groups may lead to novel functional materials, such as GeTe-Sb₂Te₃ solid solutions which have essential application prospects in the field of phase-change memory devices [27,28]. In addition, entropy stabilization provides a new direction for developing functional materials [29]. In the field of thermoelectricity, lattice disordering and distortions in high-entropy alloys (HEAs) tend to result in low thermal conductivity by enhancing phonon scattering [30–33].

In previous studies, the layer spacing indicator was proposed, which is based on the structural information of interlayer (spacing of layers between different planes perpendicular to the rotation symmetry axis) and intralayer (distance between nearby atoms on the same planes perpendicular to the rotation symmetry axis) [34]. It is effective in

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describing the structural symmetry of materials across different crystal systems and provides a quantitative descriptor of crystal structure manipulation, in order to manipulate the electronic structure as well as electrical performance in solid solutions. It is possible to realize solid solution across the crystal structures from different crystal systems, and to obtain a certain solubility in the corresponding phase region. Following this perspective, the structural similarities between I-V-VI₂ and V₂-VI₃ materials can be established [35]. The AgBiSe₂ and Bi₂Te₃ single-phase solid solutions have been successfully synthesized according to this strategy, but the poor electrical properties limit their possible application scenarios [35].

AgSbTe₂ is a typical *p*-type I-V-VI₂ semiconductor and has been considered as a promising thermoelectric material, which can be stabilized as a cubic crystal structure at room temperature. The strong anharmonicity induced by the lone pair electrons of Sb leads to extremely low lattice thermal conductivity [36,37], while the highly degenerated bands ensure the excellent electrical transport [38,39]. A high thermoelectric performance in AgSbTe2-based alloys has been frequently realized by adjusting the stoichiometric ratio of Ag/Sb [38, 40], or applying substitutions by elements such as Cd [41], Yb [42], Ti [43], and Se [44,45]. Here we applied the solid solution strategy of Bi₂Te₂-alloving, which would provide greater possibility for manipulating thermal transport properties, due to additional vacancies and defects. AgSbTe₂ is applied as a matrix and a series of solid solution alloys with Bi2Te3 are synthesized successfully. By means of this solid solution modulation with non-equiatomic anion to cation ratios, it is expected that lots of vacancies and substitutional point defects can be introduced while keeping the cubic structure skeleton unchanged, thus reducing the lattice thermal conductivity. To further improve the thermoelectric properties near room temperature, Cd was doped in the solid solution to increase the carrier concentration and thus modulate the electrical properties. The synthesized (AgSb_{0.98}Cd_{0.02}Te₂)_{0.9}(Bi₂Te₃)_{0.1} exhibits $zT \approx 0.93$ at 450 K, enabling a conversion efficiency of 4.2 % at $\Delta T \sim 160$ K for $(AgSb_{0.98}Cd_{0.02}Te_2)_{0.9}(Bi_2Te_3)_{0.1}/Co/Ag$ single-leg device. Furthermore, the fabrication of *p*-type (AgSb_{0.98}Cd_{0.02}Te₂)_{0.9}(Bi₂. Te₃)_{0.1}/n-type Ag₂Se module presents an alternative approach for recovering low-grade heat.

2. Results and discussion

The details about material synthesis, characterizations, and property measurements are given in the Supporting Information. Room-temperature powder X-ray diffraction (XRD) patterns of $(AgSbTe_2)_1_x(Bi_2Te_3)_x$ are shown in Fig. 1a. The diffraction peaks of the samples at $x \leq 0.20$ provide evidence that the materials have cubic structure, whereas impurity phase precipitation of Bi_2Te_3 starts to occur at x = 0.25, which shows that the solid solubility is saturated at this point. The

lattice constant *a* becomes larger with the increase of *x* until it stabilizes after saturation (Fig. 1b). By analyzing the crystal structures obtained by XRD Rietveld refinement (Fig. S1, Tables S1–S2), the ratio of the average layer spacing to the reference layer spacing (LS_{avg}/LS_{ref}) gradually increases with the increase of the solid solution Bi₂Te₃. This approximately linear relationship is consistent with our strategy of designing the solid solution material system.

To further validate the purity of the synthesized materials, the microstructure and composition of the samples were further characterized by scanning electron microscopy (SEM) observations and energy dispersive spectroscopy (EDS) analysis, respectively, on the samples after hot pressing. The corresponding SEM images and EDS mappings are shown in Figs. S2–S3. After exceeding the solid solubility ($x \ge 0.2$), the precipitation of heterogeneous phases is observed, which is in agreement with the XRD results. It is worth mentioning that the presence of secondary phases (silver telluride and *p*-type bismuth telluride) may affect the transport properties characterized later.

In order to reveal the thermal stability of the thermoelectric material obtained after solid solution, the results of temperature-dependent XRD for $(AgSbTe_2)_{0.9}(Bi_2Te_3)_{0.1}$ are shown in Fig. 2a. The material tends to decompose and fails to maintain a stable cubic structure at 475 K while it is completely denatured by 600 K. Detailed Rietveld refinement results of temperature-dependent XRD data are shown in Fig. S4 and Table S3. Furthermore, the resistivity of the material is tested during heating and cooling, and the maximum test temperature is sequentially increased by 20 K for each time. It changed negligibly within 460 K (Fig. 2b). The detailed characterization before and after annealing can be found in Fig. S5. These results reveal that the material properties can maintain a stable response in this temperature region. Therefore, the investigation of the transport properties of the solid solutions is focused on the temperature range of 300–450 K.

The detailed thermoelectric properties of the solid solution are shown in Fig. S6. As the solid solution content of Bi₂Te₃ increases, a growing number of vacancies are introduced into the pristine stable cubic lattice, and the lattice thermal conductivity (κ_L) basically exhibits an effective decrease due to various phonon scattering mechanisms such as dislocation scattering and interface scattering [46-49]. The sudden increase in thermal conductivity for the sample with x = 0.2 is presumably due to the Bi₂Te₃ precipitation, which has higher thermal conductivity than AgSbTe₂, in agreement with the SEM results of Fig. S2. The optical measurement (Fig. S7) shows an estimated band gap of \approx 0.44 eV for (AgSbTe₂)_{0.9}(Bi₂Te₃)_{0.1}, which is close to AgSbTe₂, exhibiting a semiconductor characteristic. The thermal conductivity of (AgSbTe₂)_{0.9}(Bi₂Te₃)_{0.1} is obviously decreased compared to the intrinsic AgSbTe2 due to the introduction of vacancies and substitutional point defects during the solid solution, where the electrical properties can be optimized by adjusting the carrier concentration, in order to further



Fig. 1. (a) Room-temperature powder X-ray diffraction (XRD) patterns, (b) composition-dependent lattice parameters, and (c) the relationship between the close-packed layer spacing ratio (LS_{avg}/LS_{ref}) for (AgSbTe₂)_{1-x}(Bi₂Te₃)_x (0 $\leq x \leq$ 0.30).



Fig. 2. (a) Temperature-dependent XRD and (b) resistivity during several thermal cycles of $(AgSbTe_2)_{0.9}(Bi_2Te_3)_{0.1}$.

improve the overall performance.

It is well known that doping with heterovalent elements is an effective method of tuning carrier concentrations. From literature reports, the doping of Cd can greatly improve the electrical and thermal properties in AgSbTe₂ [41]. As illustrated in Fig. S8, the XRD patterns of the (AgSb_{1-y}Cd_yTe₂)_{0.9}(Bi₂Te₃)_{0.1} ($0 \le y \le 0.04$) samples and the SEM-EDS mapping results demonstrate that the rock-salt cubic structure was obtained successfully. The temperature-dependent thermoelectric property after Cd doping is shown in Fig. 3. The hole concentration is increased, due to the substitution of Cd for Sb, and accordingly, the Seebeck coefficient and resistivity are significantly reduced, indicating the effectiveness of Cd doping. The Pisarenko curves and the effective

mass of the density of states m^* based on the single parabolic band model with acoustic scattering at 300 K are supplemented in Fig. S9 and Table S4. Benefiting from the intrinsically low $\kappa_{\rm L}$ and the improved electronic properties by Cd doping, the peak zT value of 0.93 is achieved at 450 K. The sample with y = 0.04 was not considered due to the presence of a weak CdTe second phase. The results show that (AgSb_{0.98}Cd_{0.02}Te₂)_{0.9}(Bi₂Te₃)_{0.1} is a promising *p*-type thermoelectric candidate for low-temperature applications.

The $(AgSb_{0.98}Cd_{0.02}Te_2)_{0.9}(Bi_2Te_3)_{0.1}/Co/Ag$ single-leg device and module were fabricated to illustrate the potential of the designed thermoelectric material for efficient and robust power generation application. Numerous studies have shown that an excellent interface is an



Fig. 3. Temperature-dependent (a) Seebeck coefficient, (b) resistivity, (c) power factor (*PF*), (d) total thermal conductivity (κ), (e) lattice thermal conductivity, and (f) thermoelectric figure-of-merit for (AgSb_{1-y}Cd_yTe₂)_{0.9}(Bi₂Te₃)_{0.1} (0 \leq y \leq 0.035), with a comparison to those of AgSbTe₂ [50] and Ag_{0.366}Sb_{0.558}Te [38].

important factor to determine the device output [6,16,51-53]. The interfacial electrical/thermal contact resistance, the bond strength and degree of interdiffusion between the barrier layer, and the thermoelectric material highly affect the efficiency and lifetime of the device. In this work, Co was chosen as the diffusion barrier layer and Ag as the electrode with (AgSb_{0.98}Cd_{0.02}Te₂)_{0.9}(Bi₂Te₃)_{0.1} based thermoelectric material. As shown in Fig. 4, Co layer is well bonded to (AgSb_{0.98}Cd_{0.02}Te₂)_{0.9}(Bi₂Te₃)_{0.1} thermoelectric material and Ag electrodes, with negligible diffusion/reaction, and the thickness of Co diffusion barrier layer is \sim 120 µm. The electrical contact resistance (R_c), including both electrode and diffusion barrier, of (AgSb_{0.98}Cd_{0.02}-Te₂)_{0.9}(Bi₂Te₃)_{0.1}/Co/Ag is estimated to be ~3.4 mΩ, and the total contact resistance at both cold and hot sides accounts for approximately 7 % of the internal resistance (R_{in}) of the single-leg device, suggesting good interfacial contacts (Fig. 4b and c). This corresponds to an average interfacial contact resistivity (ρ_c) of ~131 $\mu\Omega$ cm².

The schematics of the device performance measurement in this work are shown in Fig. S10. With a fixed cold-side temperature of ~290 K, the open circuit voltage $V_{\rm OC}$, maximum output power $P_{\rm max}$, and maximum conversion efficiency $\eta_{\rm max}$ versus temperature gradients (ΔT) for the (AgSb_{0.98}Cd_{0.02}Te₂)_{0.9}(Bi₂Te₃)_{0.1}/Co/Ag single-leg device and (AgSb_{0.98}Cd_{0.02}Te₂)_{0.9}(Bi₂Te₃)_{0.1}/Ag₂Se modules are shown in Fig. 5 and Figs. S11–S12. The slopes of the V-I curves represent the internal resistance of devices. The output power and conversion efficiency increase with current at different ΔT and reach the maximum values (Fig. 5, S11-S12). For the single-leg device, *P* and η increase with ΔT increasing from 27 K to 162 K, reaching the maximum values of ~5.6 mW and 4.2 % at $\Delta T = 162$ K (hot-side temperature $T_{\rm h} = 453$ K, coldside temperature $T_{\rm c} = 291$ K), respectively, as shown in Fig. 5a and b. Note that the conversion efficiency measurements of the (AgSb_{0.98}Cd_{0.02}Te₂)_{0.9}(Bi₂Te₃)_{0.1}/Ag₂Se modules were carried out under 380 K to avoid the phase transition in Ag₂Se at 406 K [54]. The measured voltage V, P, and η as a function of current I for two (AgSb_{0.98}Cd_{0.02}Te₂)_{0.9}(Bi₂Te₃)_{0.1}/Ag₂Se modules at different ΔT are presented in Fig. 5 and Fig. S12. It suggests that the power generation performance of the two modules coincides very well in Fig. S12. $P_{\rm max}$ and $\eta_{\rm max}$ increase as ΔT increases, $P_{\rm max}$ reaches 21 mW at the $\Delta T = 93$ K, and $\eta_{\rm max}$ is 1.8 % of the (AgSb_{0.98}Cd_{0.02}Te₂)_{0.9}(Bi₂Te₃)_{0.1}/Ag₂Se modules (Fig. 5a and b). This gives a new pairing possibility for Bi₂Te₃-free power generation modules.

The measurements of several thermal cycles within a temperature range of 290–453 K, as shown in Figs. S11m–p, indicate excellent thermal stability for the AgSbTe₂-based single-leg device. While slight decreases in $\eta_{\rm max}$ and $P_{\rm max}$ are still observed during the long-term stability measurements (Fig. 6), this phenomenon can be reasonably understood by the increased interfacial contact resistivity, since the nearly unchanged $V_{\rm oc}$ elucidates the thermal stability of the thermoelectric material at the hot-side temperature of 446 K.

3. Conclusion

In summary, the cubic $AgSbTe_2$ material was selected as the matrix to explore new *p*-type thermoelectric material candidates for near roomtemperature applications. Following the 'layer stacking indicator'



Fig. 4. (a) SEM imaging and the corresponding EDS mapping results, (b, c) line scan resistance (R) at the (AgSb_{0.98}Cd_{0.02}Te₂)_{0.9}(Bi₂Te₃)_{0.1}/Co/Ag interfaces.



Fig. 5. Power generation performance of (a–c) (AgSb_{0.98}Cd_{0.02}Te₂)_{0.9}(Bi₂Te₃)_{0.1}/Co/Ag single-leg and (d–f) (AgSb_{0.98}Cd_{0.02}Te₂)_{0.9}(Bi₂Te₃)_{0.1}/Ag₂Se module. (a, d) Output voltage *V* and output power *P*, (b, e) conversion efficiency η versus input current *I* at different temperature gradients (ΔT). (c, f) Maximum conversion efficiency η_{max} as a function of temperature gradients (ΔT). Literature results are included for comparison [5,6,15,19,20,44,50,53,55–70].



Fig. 6. Maximum conversion efficiency (η_{max}), maximum heat flow (Q_{max}), maximum output power (P_{max}), open circuit voltage (V_{OC}), internal resistance (R_{in}), and temperature gradients (ΔT) for the (AgSb_{0.98}Cd_{0.02}Te₂)_{0.9}(Bi₂Te₃)_{0.1}/ Co/Ag single-leg device during the long-term measurements.

between crystal systems, new AgSbTe₂-Bi₂Te₃ solid solutions were obtained. The thermal conductivity of the system is reduced effectively and excellent thermal stability below 450 K is validated in continuous cycling tests. By further adjusting the carrier concentration, the peak *zT* value of 0.93 was achieved in (AgSb_{0.98}Cd_{0.02}Te₂)_{0.9}(Bi₂Te₃)_{0.1}. The corresponding (AgSb_{0.98}Cd_{0.02}Te₂)_{0.9}(Bi₂Te₃)_{0.1}/Co/Ag single-leg device shows the interfacial contact resistivity of ~131 µΩ cm². The *p*-type (AgSb_{0.98}Cd_{0.02}Te₂)_{0.9}(Bi₂Te₃)_{0.1}/Co/Ag single-leg device shows a maximum conversion efficiency of 4.2 % operating at a temperature gradient ΔT of 162 K, and the *p*-(AgSb_{0.98}Cd_{0.02}Te₂)_{0.9}(Bi₂Te₃)_{0.1}/*n*-Ag₂Se module exhibits the efficiency of 1.8 % at ΔT = 93 K. This solid solution strategy provides new opportunities to search for promising thermoelectric candidates for near room-temperature applications and

to explore advanced new materials for low-grade heat recovery. More novel p/n-type thermoelectric material systems may be explored in future investigations to further expand the thermoelectric candidates.

CRediT authorship contribution statement

Di Zhang: Writing – original draft, Visualization, Validation, Methodology, Investigation, Data curation. **Min Liu:** Validation, Methodology, Data curation. **Tao Jin:** Validation, Methodology. **Long Yang:** Writing – review & editing, Visualization, Validation. **Wen Li:** Writing – review & editing, Visualization, Validation. **Yanzhong Pei:** Writing – review & editing, Visualization, Validation.

Declaration of competing interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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Appendix A. Supplementary data

Supplementary data to this article can be found online at https://doi.org/10.1016/j.mtphys.2025.101692.

Data availability

Data will be made available on request.

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